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(translated from Italian by Michele Gallinaro and Sebastian White, New York 2001)

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## **“On the Theory of Collisions between Atoms and Electrically Charged Particles”**

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When an atom, in its normal state, is illuminated with light of the appropriate frequency, it can be excited, which is to say that it goes to a quantum state of higher energy, absorbing a quantum of light. If the quantum of the exciting light is greater than the energy needed to ionize the atom, it can be ionized losing, depending on the light frequency, an electron of either the superficial layers or those deep within the atom. During the phase of relaxation of the atom there is emission of light and, depending on the circumstances, it is possible to obtain the phenomena of optical resonance or fluorescence. Some natural phenomena, very similar to these, occur also during excitation by collision.

In fact, if the atoms in a gas are bombarded with electrons of high enough velocity, they can be excited or ionized. They also can lose, if the velocity of the bombarding electrons is very large, some electrons belonging to the innermost layers of the atom.

The scope of the present work is to further specify the analogies (existing) between these two classes of phenomena, and to quantitatively derive the phenomena of excitation by collision from those of optical absorption. Therefore, we consider that when a charged particle passes near a point, it produces, at that point, a variable electric field. If we decompose this field, via a Fourier transform, into its harmonic components we find that it is equivalent to the electric field at the same point if it were struck by light with an appropriate continuous distribution of frequencies. If we now imagine that, at that point, there is an atom, the hypothesis occurs quite naturally, that the electric field of the particle produces on the atom those same effects of excitation or ionization that the equivalent light would produce.

Let us suppose that we know the absorption coefficient for light by the atom, as a function of frequency. We would then have a way to calculate the probability that a charged particle, passing with a given velocity at a given distance from an atom, will ionize it. It seems, on the other hand absolutely necessary to introduce a limitation to this correspondence between the electric field of the light and that of a charged particle. We note, in fact, that a particle of velocity,  $v$ , cannot produce phenomena, by collision, which require an energy greater than its own kinetic energy. If, instead, we perform a harmonic decomposition of the electric field it produces, we see that it will include all possible frequencies, including the highest ones. Therefore we must accept that all frequencies whose quantum,  $h\nu$ , is greater than the particle's kinetic energy can have no effect (since not enough energy is available to supply an entire quantum).

We have applied our hypothesis to three phenomena that allow experimental verification. They are:

a) Excitation of the line 2537 (a resonance line) of mercury. The optical absorption of this line is known with sufficient precision. Experiments also exist that give, at least to an order of magnitude, the probability that atoms of mercury bombarded with slow electrons are excited. Our theory gives for this probability the right order of magnitude. Naturally, given the limited precision of the experimental data, a precise test of the theory through these phenomena is not possible, but the case nonetheless is of some interest since no other theory of the excitation, by collision, of a quantum state exists, only theories of ionization by collision.

b) The number of ion pairs produced per cm of path length by  $\alpha$ -particles from Radium-C in Helium. We chose Helium because, being a monatomic gas, and possessing only k-shell electrons, it's possible to know with some precision its absorption coefficient as a function of frequency. The agreement between experiment and theory is very good. The previous theories of ionization by collision are all based on a principle different from the present theory. The essential point of

those theories can be summarized in the following way: when a charged particle passes within the vicinity of an atom it transfers to the electrons of the atom a fraction of the particle's energy. The energy transfer is calculated assuming, typically, that the electrons are free within the atom. One then allows the electron to be detached from the atom every time that the energy transferred to it via this mechanism is greater than the ionization energy of the atom.

c) The path length in Helium of  $\alpha$ 's from Radium-C. The mechanism of this phenomenon is, according to our theory, the following: the  $\alpha$  particle creates, along its path, a variable electric field whose energy is absorbed by the surrounding atoms, as if it were the electric field of a light wave. The energy, which is thus absorbed, has to obviously be subtracted from the kinetic energy of the  $\alpha$ particle which is thus decelerated. Also, in this case, the agreement between the theoretical and experimental values is excellent. Previously, the deceleration of  $\alpha$  particles in matter had been studied theoretically by Thomson and by Bohr. The former interpreted it as due to the fact that the  $\alpha$  particle, attracting electrons of the surrounding atoms along its path, transfers to them a fraction of its kinetic energy. Since, on the other hand, when one calculates this energy as if the electrons were free, one finds an infinitely large deceleration, Bohr redid the calculation assuming that the electrons inside the atoms were no longer free, but bound by quasi-elastic forces. In this way he finds a good agreement with experiment, taking for the frequency of this binding, as the case may be, optical or x-ray (Roentgen) frequencies.

In the second section of this work we establish some general formulas for the calculation of the "field of light" that produces an electric field equivalent to that of the particle.

In the third, fourth and fifth sections we discuss the application to a), b) and c).

II. Let's calculate, first of all, the spectral distributions corresponding to those of the electric field created by a particle with electric charge,  $\varepsilon$ , passing with velocity,  $v$ , at a minimum distance,  $b$ , from a point, P. The components of the electric force, at P, parallel and perpendicular to the motion of the particle, are:

$$E_1 = \frac{\varepsilon v \theta}{(b^2 + v^2 \theta^2)^{3/2}}, \quad E_2 = \frac{\varepsilon \cdot b}{(b^2 + v^2 \theta^2)^{3/2}} \quad (1)$$

Where  $\theta$  is the time-elapsd starting from the passage of the particle at the minimum distance from P. Taking now T to be a time that we will later extend to infinity, we can expand  $E_1$  and  $E_2$  in a Fourier series for all values of  $\theta$  between  $-(1/2)T$  and  $+(1/2)T$ .

We then find:

$$E_1 = \sum a_n \sin \frac{2\pi n}{T} \theta, \quad E_2 = \sum b_n \cos \frac{2\pi n}{T} \theta \quad (2)$$

where

$$a_n = \frac{2\varepsilon v}{T} \int_{-\frac{T}{2}}^{\frac{T}{2}} \frac{\theta \sin \frac{2\pi n \theta}{T} d\theta}{(b^2 + v^2 \theta^2)^{3/2}}, \quad b_n = \frac{2\varepsilon b}{T} \int_{-\frac{T}{2}}^{\frac{T}{2}} \frac{\cos \frac{2\pi n \theta}{T} d\theta}{(b^2 + v^2 \theta^2)^{3/2}} \quad (3)$$

The electric field,  $a_n \sin(2\pi n \theta / T)$  is now equal to the electric field of a light wave with intensity  $c/4\pi \times a_n^2/2$  and frequency  $\nu = n/T$ . We now let  $I(\nu)d\nu$  be the light intensity in a frequency interval  $d\nu$ , multiplied by the duration, T, of its action. We find

$$I(\nu) d\nu = \lim_{T \rightarrow \infty} \frac{cT}{4\pi} \frac{a_n^2 + b_n^2}{2} T d\nu =$$

$$= \frac{c \varepsilon^2}{2 \pi} \left\{ b^2 \left( \int_{-\infty}^{\infty} \frac{\cos 2\pi \nu \theta d\theta}{(b^2 + v^2 \theta^2)^{\frac{3}{2}}} \right)^2 + v^2 \left( \int_{-\infty}^{\infty} \frac{\theta \sin 2\pi \nu \theta d\theta}{(b^2 + v^2 \theta^2)^{\frac{3}{2}}} \right)^2 \right\} d\nu.$$

The integrals that appear in this expression can be represented through the modified Bessel functions of the second kind of order zero and one. Explicitly we have:

$$\int_{-\infty}^{\infty} \frac{\cos \omega x dx}{(1+x^2)^{\frac{3}{2}}} = 2\omega K_1(\omega)$$

$$\int_{-\infty}^{\infty} \frac{x \sin \omega x dx}{(1+x^2)^{\frac{3}{2}}} = -2\omega K_0(\omega)$$

With these formulae we immediately find

$$I(\nu) = \frac{8\pi c \varepsilon^2 \nu^2}{v^4} \left\{ K_0^2\left(\frac{2\pi\nu b}{v}\right) + K_1^2\left(\frac{2\pi\nu b}{v}\right) \right\} \quad (4)$$

where we substitute for brevity

$$B(\omega) = K_0^2(\omega) + K_1^2(\omega)$$

and we then find:

$$I(\nu) = \frac{8\pi c \varepsilon^2 \nu^2}{v^4} B\left(\frac{2\pi\nu b}{v}\right) \quad (5)$$

Since all the frequencies whose quantum is larger than the kinetic energy of the particle cannot, according to our hypothesis, have any effect, we find finally that the charged particle's passage is equivalent to the following spectral distribution

$$I(\nu) = \begin{cases} \frac{8\pi c \varepsilon^2 \nu^2}{v^4} B\left(\frac{2\pi\nu b}{v}\right), & h\nu \leq \frac{1}{2}mv^2 \\ 0 \dots \dots \dots, & h\nu > \frac{1}{2}mv^2 \end{cases} \quad (6)$$

III We now want to show, using a simple example, the application of the general method: therefore we will study the excitation by collision of spectral lines. For example, we consider an atom of mercury, and take  $\alpha(\nu)$  to be its absorption coefficient at the frequency  $\nu$ . This means that if the atom is struck by light of intensity  $I(\nu) d\nu$  it absorbs, on average, the energy  $\alpha(\nu) I(\nu) d\nu$  and therefore the absorption probability of a quantum  $h\nu$  is

$$\frac{\alpha(\nu) I(\nu) d\nu}{h\nu} \quad (7)$$

Since the absorption at the resonance line corresponds to the excitation of the atom to the 2p state, we find that (7) represents the excitation probability if only the frequencies in the interval  $d\nu$  would be effective. The probability  $\Pi(b)$  for the excitation to be caused by the passage of the particle at a distance,  $b$ , is naturally smaller than the sum of all the probabilities obtained from the individual frequency intervals  $d\nu$  when one treats these as independent. One easily finds that<sup>e</sup>

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<sup>e</sup>Effectively the probability that the atom doesn't absorb any quantum is:

$$1 - \Pi(b) = \prod \left( 1 - \frac{I(\nu) \alpha(\nu) d\nu}{h\nu} \right)$$

In which the product must be extended over all the infinitesimal intervals  $d\nu$ . Taking the logarithms of both sides we find in the limit

$$\log(1 - \Pi(b)) = \sum \log \left( 1 - \frac{I(\nu) \alpha(\nu) d\nu}{h\nu} \right) = - \int \frac{I(\nu) \alpha(\nu) d\nu}{h\nu}$$

From which we obtain equation 8.

$$\Pi(b) = 1 - e^{-\int \frac{I(\nu)\alpha(\nu)d\nu}{h\nu}} \quad (8)$$

Since, in the case of the resonance,  $\alpha(\nu)$  has a value different from zero only in a very narrow band near the resonance frequency  $\nu_o$ , we can write

$$\int \frac{I(\nu)\alpha(\nu)d\nu}{h\nu} = \frac{\alpha}{h\nu_0}I(\nu_0)$$

where we use

$$\alpha = \int \alpha(\nu)d\nu$$

Equation (8) can therefore be written, in this case as:

$$\Pi(b) = 1 - e^{-\frac{\alpha}{h\nu_0}I(\nu_0)} \quad (8')$$

To derive from this an observable directly accessible to experiment, we calculate the effective radius,  $\rho$ , of the atom for resonant excitation. That is to say the equivalent radius needed to obtain the same overall excitation probability if the excitation probability were unity within this radius. This radius is evidently given by

$$\pi\rho^2 = 2\pi \int bdb\Pi(b).$$

We therefore find, keeping in mind equation (6), that for  $\frac{1}{2}mv^2 > h\nu$

$$\begin{aligned} \rho^2 &= 2 \int_0^\infty \left\{ 1 - e^{-\frac{8\pi\varepsilon^2 c\alpha\nu_0}{hv^4} B\left(\frac{2\pi\nu_0 b}{v}\right)} \right\} bdb \\ &= \frac{v^2}{2\pi^2\nu_0^2} \int_0^\infty \left\{ 1 - e^{-\frac{8\pi\varepsilon^2 c\alpha\nu_0}{hv^4} B(x)} \right\} xdx \end{aligned} \quad (9)$$

The integral can be evaluated approximately<sup>c</sup> and one finds

<sup>b</sup>Effectively an electron, which passes with low velocity near an atom, will be strongly scattered by it. Since we are only after the order of magnitude, we'll apply equation (6) anyway.

<sup>c</sup>To find an approximate expression for :

$$I(\alpha) = \int_0^\infty \left( 1 - e^{-\alpha B(x)} \right) xdx$$

for  $\alpha \ll 1$ , we observe that when  $x < 0.4$  one has with sufficient accuracy  $B(x)=1/x^2$ . For  $\alpha < 1$  and  $x > 0.4$  one can set

$$1 - e^{-\alpha B(x)} = \alpha B(x)$$

We can then write

$$I(\alpha) = \int_0^{0.4} \left( 1 - e^{-\frac{\alpha}{x^2}} \right) xdx + \alpha \int_{0.4}^\infty B(x)dx$$

The numerical calculation gives us for the second integral  $0.973 \times \alpha$ .

The first can be calculated easily using the asymptotic expressions for the logarithmic integral and one finds that it has the value:

$$-\frac{\alpha}{2} \log \alpha - 0.705\alpha$$

We then obtain finally:

$$I(\alpha) = \left( 0.268 - \frac{1}{2} \log \alpha \right) \alpha.$$

$$\rho^2 = \frac{2\varepsilon^2 c\alpha}{\pi h\nu_0 v^2} \left( 0.54 - \log \frac{8\pi\varepsilon^2 c\alpha\nu_0}{hv^4} \right) \quad (9')$$

Experimentally the excitation by collision of the resonance line 2537 of mercury was studied by Miss Sponer<sup>d</sup>

Unfortunately this work gives us only the order of magnitude of the excitation probability. In fact, Sponer reached the conclusion that in the collisions of electrons having velocity not much greater than 4.9 volts, with atoms of mercury, inelastic collisions amount to a few percent of all collisions. To calculate  $\rho$  using (9') we take for  $v$  a velocity corresponding to a potential of 8 volts; the value of  $\alpha$  can be obtained from a work of Fuchtbauer. This author finds, in fact, that when a mercury atom is illuminated with an amount of light having the spectral distribution  $I(\nu)$ , the absorption probability of a quantum of "resonance light" is  $PI(\nu_0)$  where  $P = 8 \cdot 10^{-7}$ . Evidently  $\alpha = Ph\nu_0$ , and therefore  $\alpha = 6 \cdot 10^{-4}$ . Equation (9') gives us then

$$\rho = 0.8 \cdot 10^{-8}.$$

This value is considerably larger than that found by Sponer. If, in fact, we take the fraction of inelastic collisions to be as large as 9% we still find, for  $\rho$ , the value  $0.4 \cdot 10^{-8}$  which is about half as large as what we calculated.

On the other hand, it is easy to understand the reason for this discrepancy. When, an electron of velocity equivalent to a few volts passes at a distance of the order of magnitude of  $10^{-8}$  cms from an electron, it is already strongly scattered in such a way that its distance of closest approach is considerably larger than if it had remained on its original trajectory. One then understands that the error made in neglecting the deflection leads to an overestimate of  $\rho$ . Since, on the other hand, the uncertainty in the measurements excludes anyway a precise test of the theory, and with the present hypotheses, we obtained the right order of magnitude for  $\rho$ , it seems superfluous to attempt a more precise calculation which certainly would be more complicated.

Ionization phenomena by collision can be explained in a very similar way. It is known that all atoms beyond the limit of their principal series exhibit, both in absorption and emission, a continuous spectrum, corresponding to the transition of the valence electron to the state in which it is ionized and furthermore possesses some kinetic energy. The spectrum consists of a reasonably broad band. It has a sharp cutoff towards the red due to the limit of the principal lines and is blurred towards the violet. If, now, the velocity of the colliding electron is such that the frequency  $mv^2/2h$  lies beyond the absorption band, it is still possible, as a rough approximation, to apply equation (9') also for the calculation of the equivalent radius of the atom for ionization effects; if instead the limiting frequency falls within the absorption band the radius will naturally be smaller than found with (9') and will finally go to zero when the frequency  $mv^2/2h$  coincides with or falls below the limit of the principal series. Qualitatively this behavior of  $\rho$  as a function of  $v$  is confirmed by the experiment<sup>e</sup>. Unfortunately it seems impossible to pursue it quantitatively because precise data on the intensity of the continuous absorption spectrum are not available.

IV. Instead, it is considerably simpler to calculate the ionization produced by  $\alpha$  particles. In fact,  $\alpha$  -particles, because of their considerable mass, are hardly deflected and one can, to a good approximation, use Eq. (6). The empirical results on absorption in the region of Roentgen rays can be summarized with the following formula:

<sup>d</sup>Hertha Sponer, ZS.f. Phys. 7,p.185,1921.

<sup>e</sup>Nettleton, Proc.Nat.Acad.Amer.10,140(1924)

$$\alpha(\nu) = \frac{KZ^4}{\nu^3} + D \quad (10)$$

where  $\alpha(\nu)$  represents the atomic absorption coefficient,  $D$  is the term of the absorption due to [Compton] scattering,  $Z$  is the atomic number, and  $K$  is a coefficient, which exhibits some discontinuities at the edge of the series. For values of  $\nu$  larger than the threshold,  $\nu_o$ , of the k-series we have  $K=0.6 \times 10^{30}$ , while for  $\nu < \nu_o$ ,  $K=0.1 \times 10^{30}$ . From here one derives that the contribution to the absorption due to ionization from the k-shell, is

$$\alpha(\nu) = \begin{cases} 0, & \nu < \nu_0 \\ \frac{HZ^4}{\nu^3}, & \nu > \nu_0 \end{cases} \quad (11)$$

$H$  represents, naturally, the discontinuity of the  $K$  coefficient when we cross the threshold for the k series. One has therefore:

$$H = 0.5 \times 10^{30}.$$

The probability  $\Pi(b)$ , for an  $\alpha$ -particle to ionize the k-shell can be calculated with Eq. (8). Since the mass of the  $\alpha$ -particle is very large, we can substitute the upper limit  $mv^2/2h$ , with  $\infty$ . Thus we find:

$$\Pi(b) = 1 - e^{-\frac{8\pi c \varepsilon^2 H Z^4}{h \nu^4} \int_{\nu_0}^{\infty} \frac{d\nu}{\nu^2} B\left(\frac{2\pi\nu b}{v}\right)} \quad (12)$$

The equivalent radius can be calculated with the formula

$$\rho^2 = 2 \int_0^{\infty} b db \Pi(b)$$

We find then:

$$\begin{aligned} \rho^2 &= 2 \int_0^{\infty} b db \left\{ 1 - e^{-\frac{8\pi c \varepsilon^2 H Z^4}{h \nu^4} \int_{\nu_0}^{\infty} \frac{d\nu}{\nu^2} B\left(\frac{2\pi\nu b}{v}\right)} \right\} \\ &= \frac{v^2}{2\pi^2 \nu_0^2} \int_0^{\infty} \xi d\xi \left\{ 1 - e^{-\alpha \xi \int_{\xi}^{\infty} \frac{du}{u^2} B(u)} \right\} \end{aligned} \quad (13)$$

where for brevity one puts

$$\alpha = \frac{8\pi c \varepsilon^2 H Z^4}{h \nu_0 v^4} \quad (14)$$

The integrals in (13) can be calculated approximately<sup>f</sup> for  $\alpha \ll 1$ , and one finds

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<sup>f</sup>We need to calculate

$$I = \int_0^{\infty} \xi d\xi \left\{ 1 - e^{-\alpha \xi \int_{\xi}^{\infty} \frac{du}{u^2} B(u)} \right\}$$

When  $\alpha < 0.1$  one can break up  $I$  into an integral from 0 to 0.4 and another from 0.4 to infinity. The first can be calculated observing that for,  $\xi < 0.4$  we can substitute with reasonable approximation,

$$\rho^2 = \frac{v^2}{2\pi^2\nu_0^2} \alpha \left( 0.45 - \frac{1}{6} \log \alpha \right) \quad (13')$$

We want to apply this formula to calculate the ionization of Helium by  $\alpha$ -particles from Radium-C. Since Helium possesses k-shell electrons, and [the k-shell] is complete, (Eq. 13') will be applicable, always taking  $H = 0.5 \times 10^{30}$ . Corresponding to the ionization potential of 24.5 volts, we have  $\nu_o = 6.0 \times 10^{15}$ ; and for  $\alpha$ -particles from Radium-C,

$\nu = 1.98 \cdot 10^9$ . We therefore find  $\alpha = 0.0091$ ; with this value one finds

$$\rho^2 = 0.56 \times 10^{-16} \quad \rho = 0.75 \times 10^{-8}$$

The number of ion pairs produced per cm by the particle is naturally  $\pi\rho^2n$  where  $n = 2.6 \times 10^{19}$  represents the number of atoms per  $\text{cm}^3$  at  $15^\circ\text{C}$ .

The agreement between theory and experiment is quite satisfactory; in fact, we observe that by chance the agreement is better than could be foreseen given the precision of the data.

In fact the large uncertainty, especially in the coefficient  $H$ , could easily account for a discrepancy of 20% and perhaps more.

V. An additional experimental confirmation of our theory can be found in the treatment of deceleration of  $\alpha$ -particles in matter. We'll also apply this method to He for which, as was seen earlier, it is possible to find a reasonable value for the absorption coefficient.

First of all we want to calculate the average energy loss that an  $\alpha$ -particle undergoes passing at a distance,  $b$ , from an atom. Let  $\Pi(b)$  be the probability that the particle passing at a distance,  $b$ , ionizes the atom, and  $P(b,\nu)d\nu$  be the probability that would exist for the atom to be ionized, if only frequencies in the interval,  $d\nu$  were effective. The probability that the ionization takes place with the absorption of a quantum of frequency  $\nu$  is then:

$$\Pi(b) \times \frac{P(b,\nu) d\nu}{\int_0^\infty P(b,\nu) d\nu}$$

This ionization corresponds to an energy loss  $h\nu$ . The average energy loss of the particle is therefore:

$$\Pi(b) \times \frac{\int_0^\infty P(b,\nu) h\nu d\nu}{\int_0^\infty P(b,\nu) d\nu}$$

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$$\int_\xi^\infty \frac{du}{u^2} B(u) = \frac{1}{3\xi^2} - 0.64$$

and then the asymptotic expressions for the logarithmic integral can be applied. The second integral can instead be approximately written as

$$\alpha \int_{0,4}^\infty \xi^2 d\xi - \alpha \xi \int_\xi^\infty \frac{du}{u^2} B(u)$$

The coefficient,  $\alpha$ , calculated numerically is 0.28. Thus we find

$$I = \alpha \left( 0.45 - \frac{1}{6} \log \alpha \right).$$

This number is then 4800. Instead, experimentally, one finds 4600.

If  $n$  represents the number of atoms per unit volume, the particle passes, during path 1, within a distance of  $b$  to  $b+db$  from  $2\pi n b db$  atoms. If  $T$  is its kinetic energy, we have

$$\frac{dT}{dx} = -2\pi n \int_0^\infty b db \Pi(b) \frac{\int_0^\infty P(b, \nu) h\nu d\nu}{\int_0^\infty P(b, \nu) d\nu} \quad (15)$$

now we have

$$P(b, \nu) = \frac{I(\nu) \alpha(\nu)}{h\nu} = \begin{cases} \frac{8\pi c \varepsilon H Z^4}{h v^4} \frac{1}{\nu^2} B\left(\frac{2\pi \nu b}{\nu}\right), & \nu > \nu_0 \\ 0, & \nu < \nu_0 \end{cases}$$

formula 16 now becomes, with a simple transformation, keeping in mind Eq. (14)

$$\frac{dT}{dx} = -\frac{4c\varepsilon^2 H Z^4 n}{\nu_0^2 v^2} \int_0^\infty \xi d\xi \left\{ 1 - e^{-\alpha \xi \int_\xi^\infty \frac{du}{u^2} B(u)} \right\} \frac{\int_\xi^\infty \frac{du}{u^2} B(u)}{\alpha \xi \int_\xi^\infty \frac{du}{u^2} B(u)} \quad (16)$$

When  $\alpha$  is very small these expressions can be evaluated approximately and one finds<sup>9</sup>

$$\frac{dT}{dx} = -\frac{c\varepsilon^2 H Z^4 n}{\nu_0^2 v^2} \left( 0,96 - \log \frac{8\pi c \varepsilon^2 H Z^4}{h\nu_0 v^4} \right) \quad (17)$$

and since  $T = (1/2) m v^2$ , we find then

$$\frac{dv}{dx} = -\frac{c\varepsilon^2 H Z^4 n}{m\nu_0^2 v^3} \left( 0,96 - \log \frac{h\nu_0 v^4}{8\pi c \varepsilon^2 H Z^4} \right)$$

From this expression we can easily derive a formula which gives us the path-length travelled by the particles while their velocity is reduced from its initial value,  $\nu_0$ , to the final value,  $\nu$ . Explicitly one finds.

$$x = -\frac{m\nu_0^2}{c\varepsilon^2 H Z^4 n} \int_\nu^{\nu_0} \frac{v^3 dv}{0,96 - \log \frac{h\nu_0 v^4}{8\pi c \varepsilon^2 H Z^4}} = 2,4 \frac{m\nu_0}{nh} \int_{\frac{0,104 h\nu_0 v^4}{c\varepsilon^2 Z^4}}^{\frac{0,104 h\nu_0 \nu_0^4}{c\varepsilon^2 Z^4}} \frac{du}{\log u} \quad (18)$$

Naturally this formula is valid only when both limits are large with respect to unity, since otherwise (17'), from which we derived (18) is not applicable, and we would then need to calculate (16) exactly also for large values of  $\alpha$ . However, we can calculate the path-length in Helium

<sup>9</sup>To calculate this integral we need to use (for  $\xi < 0,4$ ) the approximate expressions.

$$\int_\xi^\infty \frac{du}{u^2} B(u) = \frac{1}{3\xi^2} - 0,64, \quad \int_\xi^\infty \frac{du}{u} B(u) = \frac{1}{2\xi^2} - 0,64$$

Furthermore it is convenient to divide the integral into a term from 0 to 0.4 which can be easily calculated using the asymptotic expressions for the logarithmic integral. To calculate the second term from 0.4 to infinity, one needs only to observe that for  $\xi > 0,4$  and  $\alpha$  very small we can take:

$$1 - e^{-\alpha \xi \int_\xi^\infty \frac{du}{u^2} B(u)} = \alpha \xi \int_\xi^\infty \frac{du}{u^2} B(u)$$

of  $\alpha$ -particles from Radium-C when their velocity is reduced by half. In fact, substituting in eqn.18  $m = 6.6 \times 10^{-24}$ ;  $\nu_o = 6.0 \times 10^{15}$ ;  $n = 2.6 \times 10^{19}$ ,  $\varepsilon = 2 \times 4.77 \times 10^{-10} = 9.54 \times 10^{-10}$ ;  $v_o = 1.98 \times 10^9$ ; we find:

$$x = 0.56 \int_{18}^{288} \frac{du}{\log u} = 32$$

Since the *range* [English word used in the original] of  $\alpha$ -particles of velocity  $v_o/2$  is equal to about  $1/8^{th}$  of the range of particles with velocity  $v_o$ , we immediately deduce for the latter 37cm.

Experimentally we find, in good agreement with this theoretical value, a *range* of approximately 33 cm.